9th Class Chemistry English Medium Online Test For Full Book

Sr	Questions	Answers Choice
1	Whcih subshells are present in L - shell?	A. S and P B. Only s -sub shell C. Only p - sub shell D. Sub shell
2	Number of electrons that an be accomodated in f - subshell	A. 6 B. 10 C. 2 D. 14
3	X is a monotic gas, which statement about this is correct	A. X burns in air B. X is coloured C. X wil displace iodine from it D. X is unreactive
4	Which is the correct formula of calcium phosphide.	A. CaP B. Ca₃P₂ C. CaP₂ D. Ca₂P₃
5	Accidents ofter result from	A. Failure to follow instructions B. Failure to use common sense C. Making mistakes D. All of the above
6	Which one of the following is not a green house effect?	A. Increasing atmospheric temperature B. Increasing sea level C. Increasing flood risks D. Increasing flood chains
7	When acids react with metals which gas is evolved?	A. O2 B. N2 C. Cl2 D. H2
8	Which group of the periodic table has the tendency to gain electrons.	A. Group -1 B. Group -17 C. Group-2 D. Group -18
9	The Earth's atmosphere is getting hotter beacuse of.	A. Increasig concentration of CO B. Increasig concentration of CO2 C. Increasig concentration of O3 D. Increasig concentration of SO2
10	The structure of C atom in diamond is.	A. Pentagonal B. Triangle C. Hexahedral D. Tetrahedral
11	Whcih one of the following is the weakest force among the atoms.	A. Intermolecular force B. Ionic force C. Metallic force D. Covelent forces
12	A bond pair is covalent molecules usually has.	A. One electron B. Two electron C. Three electron D. Four electron
13	Hydrogen bound ing is always found in	A. Non-polar molecules B. Homoatomic molecules C. Polar Molecules D. All of the above
14	Second and third periods are called	A. 1st transition series B. Normal periods C. 2nd transiion series D. 3rd transiliton serios
15	Which Isotopes is commonly used to irradiate cancer cells.	A. Cobalt -60 B. Iodine-23 C. Carbon -14

		D. Iodine-131
16	Whcih Isotopes is used for diagonosis of goiter?	A. Iodine-131 B. Cobalt -60 C. P-32 D. Sr-90
17	A complete reaction is in which	A. Only 10% reactants covert into products B. All the reactants covrt into products C. All the reactants do no covrt into products D. Half reactants covert into produts
18	Group 17 elements are called	A. Alkali Metals B. Halogens C. Noble gases D. Alkaline Earth Metals
19	Saturated hydrocarbons are also called	A. Paraffins B. Olefins C. Alkene D. Alkyne
20	Whcih element is presnet in 2nd period.	A. Lithium B. Beryllium C. Boron D. All of these