States of Matter and Phase Changes

Sr	Questions	Answers Choice
1	Solubility of whcih salt increases with the increase of temperture.	A. KNO3 B. KCL C. NaNO3 D. All of These
2	A good example of homogeneious mixture is.	A. Soil B. Ice-Creem C. Rock D. Wood
3	The study of theinteraction between drugs and biological targets, as wellas the development of new medicinal agents.	A. Inorganic Chemistry B. Organic chemistry C. Medicinal Chemistry D. Environmental chemistry
4	The simplist form of matter	A. Liquid B. Solid C. Gas D. Both b and c
5	Which of the following is a heterogeneous mixture	A. Concrete mixture B. a solution of potassium nitrate in water C. a solution of calcium hydroxide in water D. Hot chocolate
6	Which branch of chemistry tells us sulphuric acid is extremely corrosive to skin.	A. Organic chemistry B. Physical chemistry C. Inorganic chemistry D. Biochemistry
7	Whcih one of the following wil show negligible effect of termperture on its solubility.	A. KCl B. KNO3 C. NaCl D. NaNO3
8	Any thing that has mass and occupies space is called.	A. Liquid B. Gas C. Solid D. Matter
9	Matter is present in neon signs in the state of	A. Supereritical fluid B. Gas C. Liquid Crystal D. Plasma
10	Which one of the following is heterogeneious mixture.	A. Milk B. Ink C. Sugar solution D. Milk of magnesia
11	Building block of all matter is.	A. element B. Compound C. Mixture D. Atom
12	The branch of chemistry that helps to treat dseases and synthesize new medicines.	A. Physical Chemistry B. Environmental Chemistry C. Organic chemistry D. Inorganic chemistry
13	A state of matter whose properties are between those of liqids and crystalline solids.	A. Plasma B. dark matter C. Liquid crystal D. Supercriticl fluid
14	Mixture can be separated by means of	A. Nuclear method B. Phyical medhod C. Chemical Method D. All of these
15	Which is deal with compositon, structure, properties, behavior and changes of matter and energy.	A. Technology B. Chemistry C. Engineering

		D. Science
16	Matter can be described by both is	A. Chemical Properties B. Physical Properties C. Physical propeties and chemcial properties D. None of these
17	How many state of matter exist.	A. One B. Two C. Three D. Four
18	A substance formed when two or more different elements combine chemically.	A. Compund B. Solution C. Element D. Atom
19	In diamond, each C-atom is covaently bonded wiht.....C-atom	A. Two B. Three C. Four D. Five
20	Diamond is.....	A. conductor B. non -conductor C. both a and b D. None of these