Group Properties and Elements

Sr	Questions	Answers Choice
1	Non-metal form	A. Positive ion B. Neurtal atom C. Negative ion D. None of these
2	Sodium is extremly reactive metal but it does not react with.	A. Hydrogen B. Sulphur C. Nitrogen D. Phosphorous
3	Which one is the most reactive metal.	A. Gold B. Sodium C. Potassium D. All of these
4	In which element there exists the strongest foreces of attraction beween atoms.	A. Mg B. Ca C. Ba D. Sr
5	The following solutions of a halogen and a sodium halide are mixed together Which solution will turn dark becasue of a reactions.	A. Br2 and NaF B. Br2 and NaCl C. Cl and NaF D. Cl2 and NaI
6	Which metal exists in liquid form at room temperature.	A. Sodium B. Potassium C. Mercury D. None of these
7	Group 17 elements are called	A. Alkali Metals B. Halogens C. Noble gases D. Alkaline Earth Metals
8	Which compunddo you expet the be coloured.	A. KCl B. NiCl2 C. BaCl2 D. Al Cl2
9	Non-Metals are of heat and electricity	A. Good conductor B. Moderate conductor C. Bad Conductor D. None of these
10	Platinu alloyed with whih metal is used as catalyst in automobiles an catalytic convertor.	A. Palladium B. Rhodium C. Gold D. Both a and b
11	Which element has high electronegativetivity Value.	A. Chlorine B. Oxygen C. Fluorine D. Nitrogen
12	Which of the followig is less malleable	A. Gold B. Silver C. Sodium D. Iron
13	The non -metal which is present in group -15 of periodic table.	A. Nitrogen B. Iodine C. Carbon D. Sulphur
14	Metals can form ions carrying charges.	A. Uni Positive B. Di positive C. Tri positive D. All of these
15	Metals possess	A. Ionic bond B. Metallic bond C. Covalent bond D. Co ordinate bond

16	Which is weak acid	A. HF B. HBr C. HI D. HCl
17	Metals lose their electrons easily beacuse.	A. Good conductors of heat B. They are electropositive C. The have electron affinity D. They are electronegative
18	Which one of the following is the lightest metal	A. Lithium B. Calcium C. Magnisium D. Sodium
19	Non metallic oxides are.	A. Basic in nature B. Acidic in nature C. Amphoteric D. All of these
20	Which oxide is themost basic oxide.	A. Na2O B. MgO C. Li2O D. CO