Periodic Table and Periodic Properties

Sr	Questions	Answers Choice
1	Noble gases are present in 18 group also called	A. I A B. II B C. VIII A D. VIII B
2	Group 16 elements are also called.	A. Chalocogens B. Pncitogens C. Alkali metals D. Haloges
3	Which is the correct general trend in the variation of electron affinity in a group	A. It becomes less negative from top to bottom B. It becomes more negative from top to bottom C. It remains the same D. It has no definit trend and changes irregularly
4	Select basic oxide.	A. MgO B. CO2 C. SiO2 D. Al2 O2
5	The term "Halogen" mean	A. Ash former B. Salt Former C. Coper giver D. Sulpher giver
6	Metallic oxides are.....in nature.	A. Acidic B. Basic C. Amphotric D. Depends upon oxide
7	2nd Value of electron affinity	A. Will be negaive always B. Will always be positive C. Equal to 1dy value D. Double than 1st Value
8	Concept of Triads was given by	A. Mendeleev B. L. Meyer C. Dobereiner D. Newlands
9	Which one do you think is correct about metallic character.	A. It decrease frin tio ti vittin ub a griyo B. It increases from top to bottom in a group C. It remians constant fro left to right in a period D. It increases from left to right in a period
10	Which property increases as you go down a group in the periodic table?	A. Atomic radium B. Elecron affinity C. Electronegativity D. Ionization Energy
11	Which one of the following element has the highest ionization energy.	A. Argon B. Sodium C. Aluminium D. Magnesium
12	Electronegativity value of "Cs" is	A. 2.1 B. 0.79 C. 0.1 D. 4
13	Select metalloid from the following.	A. Po B. Cl C. Fe D. P
14	Recognize the elemetn if it has 3 electron shells. belongs to 's" block and has 2 electrons in its outer most shell	A. Calcium B. Sodium C. Magnesium

		D. Potassium
15	John Newlands gavt he idea of.	A. Law of triads B. Law of Octaves C. Modern Periodic law D. Curves between weight and volume
16	Acidic oxides are	A. Metallic oxides B. Non Metallic oxides C. Metalloid oxides D. All
17	Which is the correct trend in variation of electrongativity along a period of the periodic table.	A. It tdecreases from left to right across a period B. It increases from left to right across a period C. It remians constant D. It has no definite trend
18	Select which is not alkaline earth metal	A. Ba B. Be C. Sr D. CO
19	Element that can show variable oxidation states.	A. Mg B. Ca C. S D. Mg
20	What is teh oxidation state of sulfur in the sulfate ion $SO^{-2}4$	A. +4 B. +6 C. +2 D. 0