Chemistry Fsc Part 2 Chapter 11 Online Test

Sr	Questions	Answers Choice
1	Phenol after reduction with hydrogen changes to	A. Picric acid B. Benzene C. Cyclohexane D. Cyclohexanol
2	Which compound shows maximum hydrogen bonding with water	A. CH₃OH B. C₂H₅OH C. CH₃- O - CH₃ D. C₆H₅OH
3	Methyl alcohol can be distinguished from ethyl alcohol by	A. Action of Cl2 B. Action of NH3 C. Dissolving in H2O D. NaOH +I2
4	According to Lewis concept ethers behave as	A. Acid B. Base C. Acid as well as a base D. Noe of them
5	Ethanol on oxidation in the presence of K2Cr2O7 and Conc. H2SO4 changes to.	A. Acetaldehyde B. Ethane C. Ethene D. CO2 and H2O
6	Phenol is also known as	A. Citric acid B. Carbonic acid C. Carbolic acid D. Maleic acid
7	The conversion of ethene to ethanol is an example of.	A. Hydration B. Dehydration C. Neutralization D. Esterification
8	---------- compound shows extensive hydrogen bonding with water	A. C₂H₆ B. H₂S C. C₂H₅OH D. CH₃Cl
9	According to Lewis concept, ethers behave as	A. Acid B. Base C. Nucleophile D. Solvent
10	The IUPAC name of CH3OCH6H5 is	A. Methyl phenyl ether B. Methaoxy benzene C. Phenoxy methane D. methoxy phenyl
11	When ethyl alcohol is heated, with NH3 in presence of ThO2 then	A. O-H bond is broken B. C-O bond is broken C. Ethene is formed D. Ethane is formed
12	Phenol can be prepared from chlorobenzene by	A. Williamson synthesis B. Down's process C. Kolbe reaction D. Cannizzaro reaction
13	Alcohol obtained by fermentation is only upto	A. 10% B. 12% C. 20% D. 95%
14	The product of fermentation of sucrose is	A. Ethanol and H2O B. Ethanol and CO C. Ethanol and CO2 D. Glucose and CO2
15	The conversion of ethanol to ethene is an example of.	A. Dehydration B. Hydration C. Hydrogenation

		D. Fermentation
16	Which compound is more soluble in water	A. C₂H₅OH B. C₆H₅OH C. CH₃COCH₃ D. n - hexanol
17	Which one of the following methods is used for the preparation of ether.	A. Kolbe's reaction B. Frankland reaction C. Williamson synthesis D. Down's process
18	Which compound is called universal solvent	A. CH₃OH B. C₂H₅OH C. CH₃O CH₃ D. H₂O
19	Ethanol reacts with Na metal to form sodium ethoxide. What product will be formed when C2H5ONa reacts with methyl bromide.	A. C2H5OC2H5 B. C2H5OCH3 C. CH3COC2H5 D. C2H5Br and NaBr
20	Which compound shows maximum hydrogen bonding with water.	A. CH3OH B. C2H5OH C. CH3-O-CH3 D. C6H5OH