Chemistry Fsc Part 2 Chapter 2 Online Test

Sr	Questions	Answers Choice
1	The ore CaSO4.2H2O has the general name.	A. Gypsum B. dolomite C. Calcite D. Epsom salt
2	Which meal is least reactive	A. Li B. NaOH C. K D. Pb
3	Nelson's cell is used to prepare	A. NaOH B. Na₂CO₃ C. Na metal D. NaCl
4	The most metallic elements from the following is	A. Arsenic B. Oxygen C. Antimony D. Bismuth
5	Which substance is used to lower the melting point of NaCl in Down's cell	A. CaCO3 B. CaCl2 C. MgSO4 D. AlCl3
6	Which elements are deposited at the cathode during the electrolysis of brine in diaphragm cell	A. H₂ B. Ba C. Ra D. Rn
7	Which on eof the following alkaline earth metal form peroxide when heated with oxygen at 600 ^{o}C	A. Magnesium B. Calcium C. Strontium D. Barium
8	Down's cell is used to prepare.	A. Sodium carbonate B. Sodium bicarbonate C. sodium metal D. Sodium hydroxide
9	Which element reacts with alkali and H2 gas is produced.	A. Be B. Mg C. Ca D. Sr
10	Which of the following sulphates is not soluble in water.	A. Sodium sulphate B. Potassium sulphate C. Zinc sulphate D. Barium Sulphate
11	Sodium is collected over molten sodium chloride in Down's cell, because.	A. It does not react with NaCl B. It is lighter in weight and floats on molten NaCl C. Its reactivity is greater than NaCl D. None of these
12	Which one is not use of lime	A. Extraction of metals B. Ceramic industry C. Preparation of bleaching powder D. Preparation of gypsum
13	The mineral $CaSO_4.2H_2O$ has the general name	A. Dolomite B. Gypsum C. Calcite D. Epsom Salt
14	Which is insoluble in water.	A. BeSO4 B. MgSO4 C. CaSO4 D. BaSO4
15	Dolomite is a carbonate of	A. Be B. Mg C. Na

		D. Ba
16	The oxides of beryllium are	A. Basicity B. Rancidity C. Amphoteric D. none of all
17	Chile saltpeter has the chemical formula	A. $NaNO_3$ B. KNO_3 C. $Na_2B_4O_7$ D. $Na_2CO_3.H_2O$
18	Which element reacts with carbon to form metal carbide.	A. Li B. Na C. K D. All of these
19	The crystals of caustic soda, NaOH, are hygroscopic when these crystals are exposed to air.	A. Gain water and remains solid B. Gain water and becomes liquid C. Lose mass and remain solid D. Remain unchanged on heating.
20	Which is more basic	A. RbOH B. NaOH C. KOH D. Li OH